# **Electronic Band Structure of ALD MoTe<sub>2</sub>/TiO<sub>2</sub>** Heterostructures

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## Background

**Solar Cell Operations** 

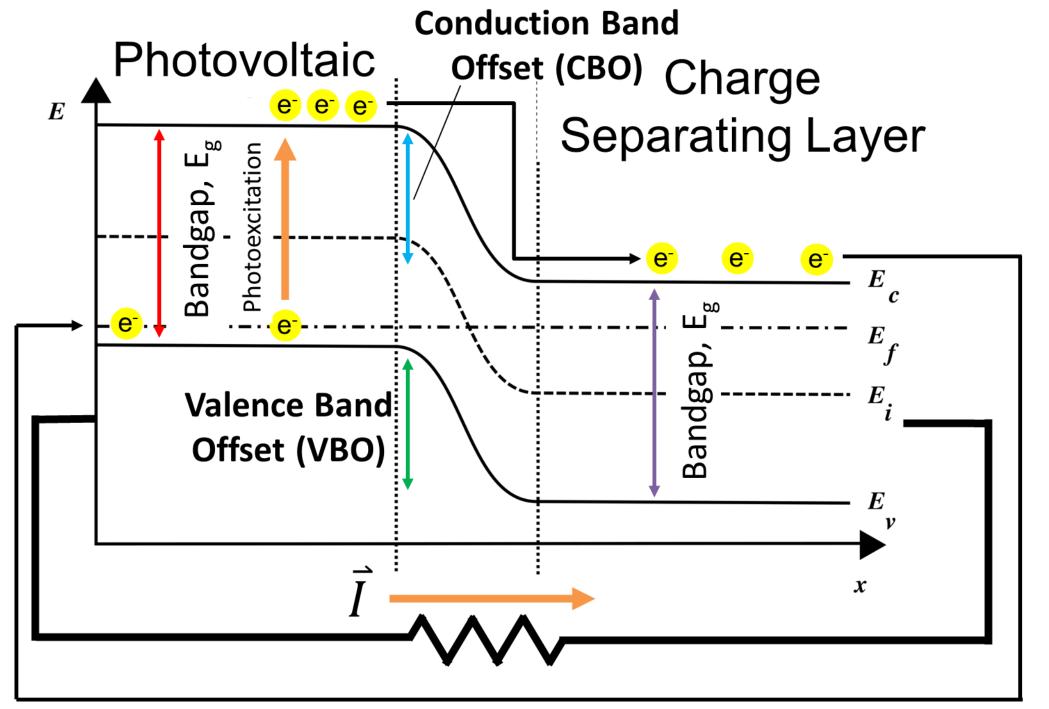
A solar cell absorbs photons to excite electrons, then pass these through a circuit to generate electrical power. This requires a photovoltaic and a

## This work

<u>Characterizing the chemical and electronic properties of ALD MoTe<sub>2</sub>/TiO<sub>2</sub></u> heterostructures

Observe alignment of  $E_V$  and  $E_C$ , Observe chemical effects at the interface between MoTe<sub>2</sub> and TiO<sub>2</sub>

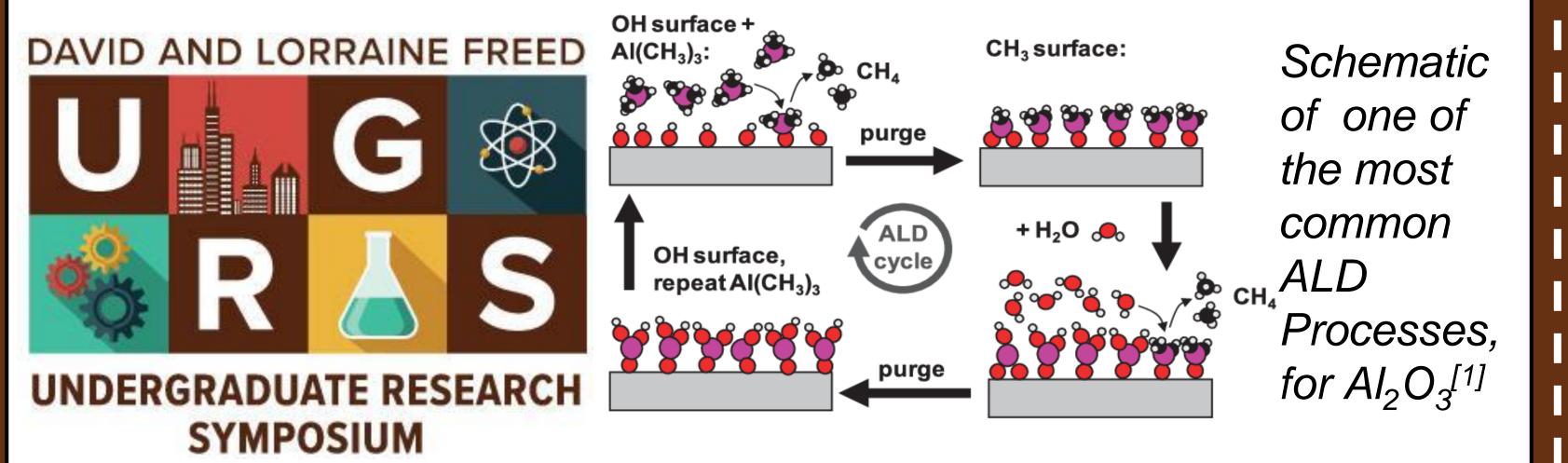
layer to separate out excited electrons before they decay.



The diagram to the left shows the alignment of the valence bands (EV, valence electrons' "ground state") and conduction bands (EC, first excited state) in such a device. Important parameters for the operation of the device are labeled.

#### **Atomic Layer Deposition**

In Atomic Layer Deposition (ALD), a substrate is alternately exposed to lowvacuum pressures of two precursors, which react to form a desired surface species in a self-limiting way. Films can be made as thin as one atom



#### MoTe<sub>2</sub>

MoTe<sub>2</sub> is a promising photovoltaic which very strongly absorbs visible light (films a few nanometers thick appear black). It also has unique properties as a 2D material. A single atomic layer of MoTe<sub>2</sub> has a direct  $E_{\alpha} = ~1.15 \text{ eV}$ , while thicker films have an indirect  $E_{\alpha} = ~1.00 \text{ eV}^{[2]}$ 

#### Samples

## 4nm amorphous TiO

1.7-5.6nm MoTe<sub>2</sub>

## Al<sub>2</sub>O<sub>3</sub> Substrate

Heterostructures with and without  $TiO_2$  at a range of MoTe<sub>2</sub> thicknesses were prepared and analyzed using XPS and UV-VIS Spectroscopy

#### TiO<sub>2</sub>

 $TiO_2$  is a transparent insulating oxide with  $E_{a} = ~3.2 \text{ eV}$  used as an electron collecting layer for MoTe<sub>2</sub>. Amorphous TiO<sub>2</sub> can be deposited using a well studied and effective ALD process<sup>[3]</sup>

- Deposited 89 ALD cycles of TiO<sub>2</sub>
- Constant thickness of 4nm used<sup>[4]</sup>
- 15, 25, 50, 80, or 120 cycles of MoO<sub>x</sub> deposited via ALD
- Annealed in Te vapor atmosphere at 500°C<sup>[2]</sup>

Commercially Available singlecrystal sapphire substrate

## **Future Work**

- Repeat experiments on more samples to confirm findings
- Knowledge of Energy levels in MoTe<sub>2</sub>/TiO<sub>2</sub>

heterostructures will be used to inform further study of excitons in MoTe<sub>2</sub>

 Applying Transient Absorption Spectroscopy, a pump-probe method identifying energy levels and lifetimes of excited states

## -X-ray Photoelectron Spectroscopy (XPS)

 $(E_{CL}^{B} - E_{v}^{B})$ 

(ECL - EV

AI KQ 1486.701  $\mathsf{BE} = \mathsf{E}_{\mathsf{incident}} - \mathsf{E}_{\mathsf{kinetic}} - \mathsf{\Phi}$ 

Measures e<sup>-</sup> binding energy (BE)

XPS spectra taken of Mo3d, Te3d, Te4d, VB edge, and Ti2p on post deposition samples, different peaks for unique chemical species Can identify  $E_V$  energy relative to core levels Find V<sub>B</sub> Edges via the Kraut method:

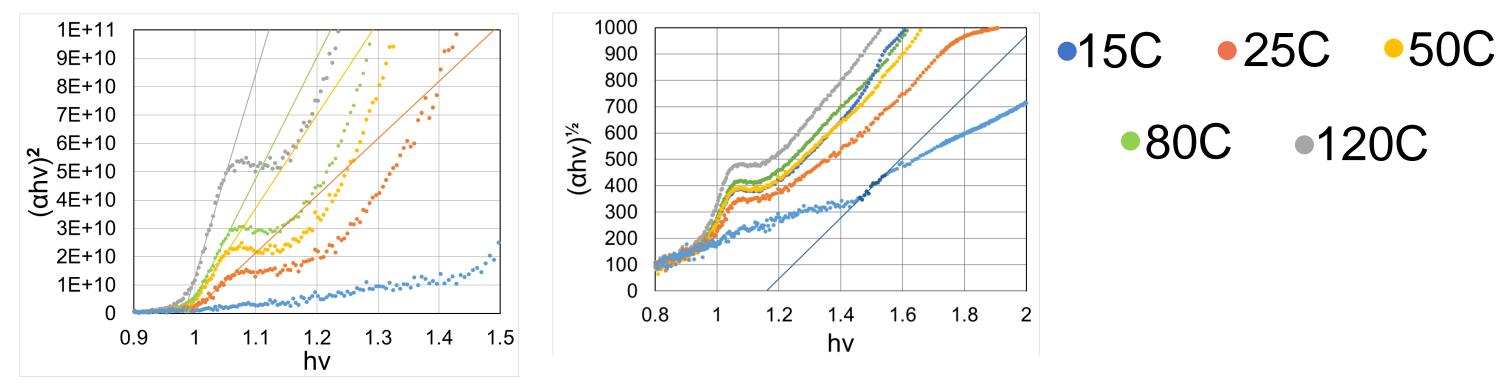
> $\Delta E_{B}$ ,  $E_{CI}^{B} - E_{V}^{B}$ , and  $E_{CI}^{B} - E_{V}^{B}$  measured directly,

 $\Delta E_V$  follows from these

 $\Delta E_{c}$  follows from  $\Delta E_{v}$  and  $E_{a}$ 

## -UV-Visible Spectroscopy (XPS)

Measures absorption coefficient as function of wavelength Performed on all thicknesses of bare MoTe<sub>2</sub> samples



- Tauc Plots used to identify bandgap: characteristic linear features appear for hv vs different powers of  $\alpha$ hv, extrapolating to y=0 gives bandgap
  - Direct  $E_a = 1.15$  eV seen for thinnest sample, indirect  $E_a = 0.99$  eV for others
    - Consistent with typical behavior of MoTe<sub>2</sub> as a 2D material

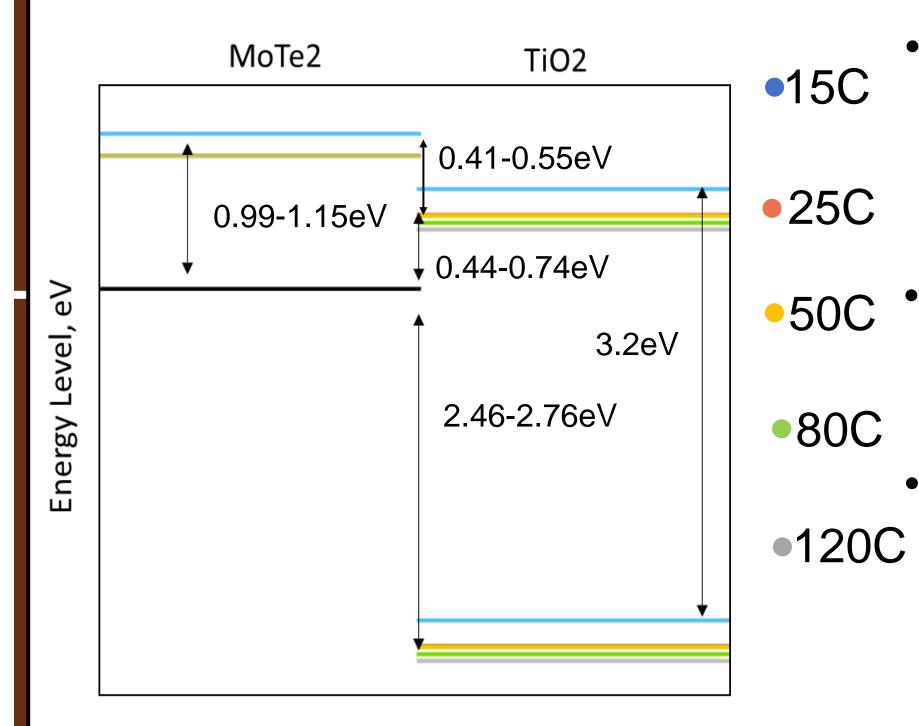
MoTe2

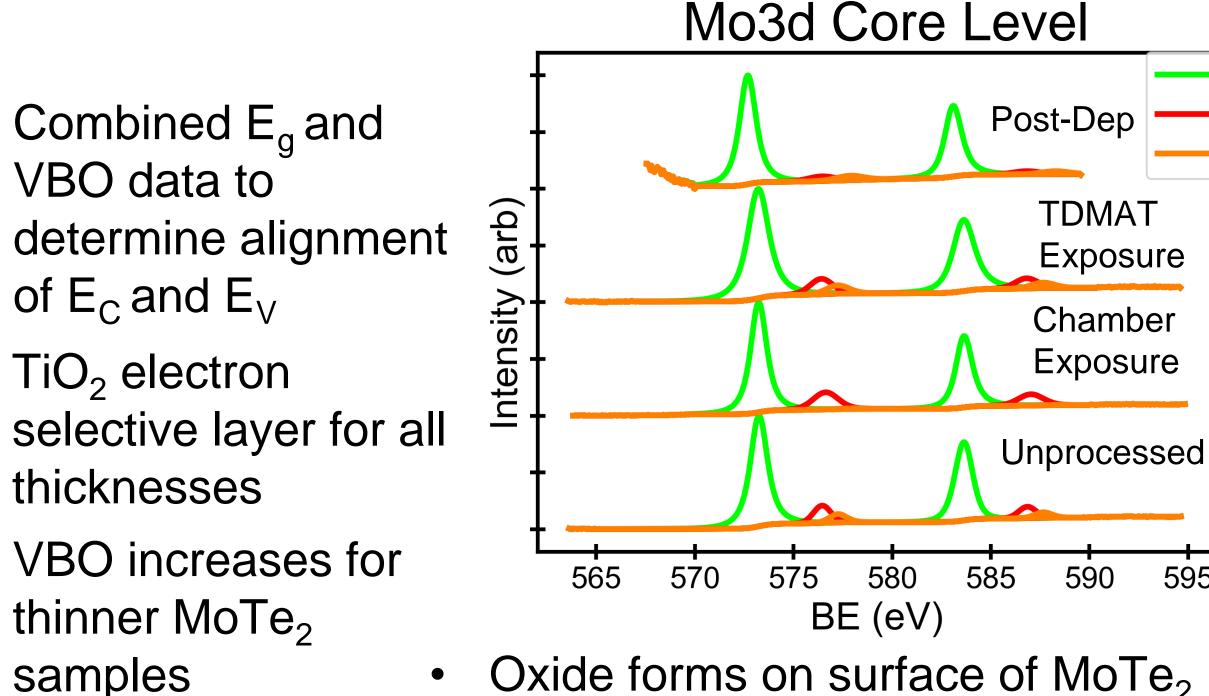
TeO2

TeO3

595

#### Results





- Oxide forms on surface of MoTe<sub>2</sub>
- Exposure to TiO<sub>2</sub> precursor found to reduce oxidation

#### References

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